## **KEY ~ CHEMISTRY ~ PERIODIC TRENDS PRACTICE #2**

]	l) Ne
2	2) Mg
3	3) chlorine atom
۷	1) aluminum ion
XX71-1-1-1	1-1:1-1-4-1-4-1-4-1-4-1-4-9
	he highest electronegativity?
	5) O
6	6) V
7	7) Br
8	B) Si
Which has the <i>lowest ionization energy</i> ?	
	0.
	9) Ga
1	10) Pt
1	11) Cf
1	12) Ra
Which has the <i>highest electron affinity</i> ?	
	13) Br
	14) N
1	15) Li
1	16) Po

17-30 answers vary

Which has the *smallest atomic size* (radius)?